

## Chem 1105 Experiment 7 Answers

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### Chem 1105 Experiment 7 Answers

CHEM 1105 Experiment 7 2 SAMPLE CALCULATIONS EXAMPLE 1. How many grams of iron(III) oxide (ferric oxide),  $\text{Fe}_2\text{O}_3$ , are formed from the reaction of 5.00 g of iron metal with excess oxygen gas? Step 1: Balanced reaction.  $4\text{Fe (s)} + 3\text{O}_2\text{(g)} \rightarrow 2\text{Fe}_2\text{O}_3\text{(s)}$  Step 2: Moles of "given" substance. The given substance is the iron metal, since its amount is given.

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